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# Improvement of Epitaxial Silicon Solar Cells on Upgraded Metallurgical Silicon by Heat Treatment.

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#### Improvement Of Epitaxial Silicon Solar Calls On Upgraded Metallurgical Silicon By Heat Treatment تحصين الخلايا الشمسية صن الصيليكون علي صيلبكون متالورجي معـــ بواسطة المعالجة الحرارية<br>Roshdy A. Abderrassoul, Electronic Eng. Dept., Mansoura University, Mansoura, Egypt.

الخلاصة لـ تم تطيد سيلبكون متالورجي معدل باستخدام الاستخلاص بواسطة المــاء ، مصرعات الله عن الله الله الله الله الله عن الله الله عن الله الله عن الله الله عن الله عن الله عن الله عن ال<br>العلكى فى اتحاه واحد على الجرافيت ليعطى طبقات تحتية ذات مناحة كبير الله عن الله عن الله عن الله عن الله عن الل آ) بترس*يب* وطة موجب ـ بالب على الطبقات التحتية بواسطة الاختزال الحرارى لشالث كلا سورو السيلان ، وكانت كغاءة التحويل الجفد ـ مُوسَّى لـها عند كَتلة هُواءَ (( ك، ه، ١ أ ) 

Abstract : Metallurgical - grade silicon upgraded by aqua regia extraction was unidirectionally solidified on<br>graphite to yield large - area large - grain substrates. The concentration of impurities in the silicon substrates was measured using the atomic absorption technique. Epitaxial silicon layers were deposited on the metallurgical silicon substrate by the thermal reduction of trichlorosilane. Heat<br>treatment in an inert atmosphere was found to improve the<br>characteristics of the solar cells, presumably due to the<br>diffusion of metallic impurities in the substr epitaxial layers to the grain boundaries. Large - area (more than  $30$  cm<sup>2</sup>) solar cells, having an air mass one (AM1)<br>photovoltaic conversion efficiency of 9 %, were prepared by depositing a  $p - n$  junction epitaxially onto the<br>metallurgical silicon substrate. The solar cells were<br>characterized by darkcurrent - voltage, illuminated current - voltage, spectral response and minority carrier diffusion length measurements, and SEM and EBIC (electron - beam induced current) techniques. The effect of heat treatment on these parameters was studied.

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#### I. Introduction :

Metallurgical - grade silicon having a purity  $\alpha$  f about 98% is an attractive starting material for low cost silicon solar cells. The use of metallurgical silicon for solar cell fabrication has been studied by several investigators [1-4]. Epitaxial silicon layers of controlled thickness and dopant concentration deposited on metallurgical silicon substrates have been shown to be a promising photovoltaic material [5]. Impurities in metallurgical silicon must be removed as completely  $a5$ possible using low cost techniques, since most metallic impurities, particularly titanium, vanadium, and zirconium, have been shown to siginificantly reduce the minority carier lifetime in single crystal silicon [6,7]. Furthermore, the metallurgical silicon substrate must have relatively large grains, since silicon is an indirect bandgap semiconductor with a gradual absorption edge, and since large grains are required to achieve reasonable photovoltaic collection efficiencies.

Metallurgical silicon contains iron and aluminum as the major impurities at concentrations of a few tenths of<br>a per cent; other impurities include boron, chromium, copper, manganese, nickel, titanium, vanadium. zirconium,.., etc., at concentrations up to several hundred parts per million. The purification of metalsilicon in the semiconductor industry involves luroical the trichlorosilane intermediate, and is energy intensive and cost ineffective.

Several direct purification techniques, such as acid extraction [8], gettering [9], and the treatment of the melt with reactive gases [10,11], have been reported. Although none of these techniques is completely effective, each has its own merit. In this paper, acid extraction was used for the partial purification of metallurgical silicon because of its simplicity and low energy requirement. The partially purified material was made into a large grain substrate by melting and unidirectional solidification on graphite. Thermal reduction of trichlorosilane by hydrogen containing appropriate dopants was used for the epitaxial deposition of the active layers of the solar cells. Heat treatment was found to significantly enhance thesolar cell characteristics, as shown by measuring the photovoltaic

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conversion characteristics of the cells and by the<br>electron – beam induced current (EBIC) technique, presumably due to the diffusion of the metallic impurities in the substrate and in the epitaxial layer to the grain boundaries.

### II. Acid Extraction of metallurgical silicon :

Many metallic impurities such as iron, aluminum, copper, gold, nickel, titanium, .. etc., are highly<br>soluble in molten silicon, while their solubilities in solid silicon are relatively low [12]. Thus, during the solidification of molten metallurgical silicon during the manufacturing process, the majority of these impurities should precipitate at grain boundaries or at interstitial positions, presumably in the form of silicides or other<br>compounds. The pulverization of the cast metallurgical silicon generally takes place at grain boundaries and the<br>small grains (1 - 2 mm, for example) are either single crystals or aggregates of two or three crystals [1].<br>Therefore the majority of the metallic impurities are<br>concentrated at the surface of the grains, and treatment of the pulverized metallurgical silicon with acids should be effective in the removal of these impurities, as well as the impurities inside the grains. However, the<br>dissolution of the impurities was found to be a slow process and impurities within the grains cannotbe removed. Table (1) shows the concentration of impurities in as - received pulverized metallurgical grade silicon,<br>determined by atomic absorption analysis. Aqua regia was found to be more effective than hydrochloric acid or

sulfuric acid in the extraction of metallic impurities

from pulverized metallurgical silicon [8].

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analysis.

Element Concentration (at. ppm) As-received metal-Metallurgical silicon lurgical silicon after aqua - regia treatment  $A<sub>1</sub>$  $610$ 320 Cu  $10$ 5 Fe 1500 350 Mn 4<sub>m</sub> 10 25 Ni 15

Table (1). Concentration of impurities in as - received<br>metallurgical silicon and metallurgical silicon purified by aqua - regia extraction determined by atomic absorption

The conecntration of iron in pulverized metallurgical silicon after aqua regia treatment was found by the atomic absorption analysis to have reduced to 350 (at. ppm) [13]. It is expected that concentrations of other metallic harmful impurities, such as aluminum, would also be significantlly reduced after acid extraction.

Acid - treated pulverized metallurgical silicon was uniformly spread on the top of a 28 cm  $\frac{1}{x}$  7.5 cm x 1.5 cm graphite plate, the surface of which was roughened by sandblasting to increase the  $Si -$  substrate interfacial tension. The graphite plate was loaded into a fused silica tube of an inside diameter of 10.5 cm in a helium atmosphere, and was rf heated by induction using a 450 kHz rf source. The temperature profile along thegraphite plate was controlled by adjusting the spacings between the turns of the rf coil so as to yield a temperature gradient of 80 - 100 °C along the length of the substrate. The entire silicon charge was first melted, then the rf power was<br>reduced so that unidirectional solidification started from one and of the substrate to the other. The solidification

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was initiated from the surface of the melt and proceeded inwards, in order to minimize the crystallographic orienting effects of polycrystalline graphite. The thermal conditions were controlled so as to yield a planar solid liquid interface during the solidification process, resulting in elongated crystallites, several centimeters long. The rate of solidification is the most important parameter affecting the morphology of the unidirectionally solidified material. High solidification rates (e.g. 5<br>cm/min) resulted in non - planar surfaces, with many<br>ridges, faces, and valleys. Lower solidification rates<br>(e.g. 1 - 2 cm/min) resulted in planar surfaces. X-ray diffraction examinations of the planar material indicate that the majority of the crystallites are of (110) orientation. Grain boundaries, multiple twins, and dislocations are the major crystallographic defects in the unidirectionally - solidified metallurgical silicon substrates.

### III. Heat treatment of metallurgical silicon substrates :

Metallurgical silicon substrates were subjected to heat treatment in an inert atmosphere, such as helium, at 700 °C for various durations. In all cases the concentration of metallic impurities at grain boundaries was found to increase appreciably [14]. Thus the grain boundaries serve as a sink for the precipitation of metallic impurities, and the presence of grain boundaries may be considered as an advantage when using chemically upgraded metallurgical silicon for solar cell purposes [15]. Segregated impurities at grain boundaries may affect trapping and surface recombination; however, such effects will not appreciably alter the extent of carrier<br>recombination at grain boundaries and therefore the photocurrent and photovoltage will not be changed.

IV. Epitaxial silicon solar cells :

Epitaxial silicon films were deposited on silicon

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substrates, prepared by the unidirectional solidification / of aqua - regia treated metallurgical silicon on graphite, using the thermal reduction of trichlorosilane  $h\nu$ hydrogen. The substrate was placed in a fused silica tube of inside diameter 10.5 cm and heated externally using an r.f. generator. In situ etching of the substrate with a mixture of hydrogen and hydrogen chloride at 1200 °C was used to remove 2 - 3 um of silicon from the substrate<br>surface. Silicon was then epitaxially deposited at 1100°C. The substrate is always p type having a resistivity of0.04  $-$  0.06  $\Omega$ .cm. Successive deposition of  $p-$  and  $n-$  type silicon films was used to produce a thin film solar cell. The p layer had a resistivity of  $0.2 - 0.5$   $\Omega$ .cm and was 20 - 25 µm thick, while the n layer was 2 µm thick and had a<br>resistivity graded from 2 x 10<sup>-4</sup> to 4 x 10<sup>-3</sup> Q.cm, After the deposition the solar cells were heated in helium at700 C to allow the diffusion of metallic impurities to grain boundaries. The grid contact was then applied to the front surface of the cell by evaporatring 1000  $\AA$  of titanium<br>followed by  $2 - 3$  µm of silver through a metal mask, and then annealing in hydrogen at 500 °C. The graphite plate served as the back contact. Tin oxide antireflective (AR) coating was applied by the oxidation of tetramethyltin in an argon atmosphere.

The solar cells were characterizerd by current  $$ voltage and spectral response measurements. Figure 1 shows the current - voltage characteristics of a thin film silicon solar cell of area 37 cm<sup>2</sup> under illumination with quartz - halogen lamps at air mass one (AM1) conditions. The open - circuit voltage,  $V_{a,c}$ , short - circuit current<br>density,  $J_{a,c}$ , and fillfactor FF are 0.57 V, 20.8 mA/cm<sup>2</sup>, and 75.6%, respectively, corresponding to anAM1 efficiency<br>of 8.95%. The heat treatment was found to increase the<br>photovoltage and photocurrent by up to 5% and 10%, respectively, and thus to increase the photovoltaic conversion efficiency by up to 20%.

The spectral response of the cell was measured at a set of discrete wavelengths selected using interference filters, and a calibrated single - crystal silicon solar cell of area 4 cm<sup>2</sup> was used as a reference. Figure 2 shows the spectral response ofthe same solar cell, compared with that of the calibrated cell. The peak response of the polycrystalline cell occurs at a shorter wavelength than that of the single crystal cell. In addition, the quantum

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efficiency in the polycrystalline cell is lower than that in the single crystal cell. The shift in the peak response and the lower quantum efficiency are not unexpected inview of the small thickness of the active region of the polycrystalline cell and the short minority carrier diffusion length in the cell.

#### V. Results :

Figure 3.a shows an SEM picture of a non annealed silicon solar cell, and figure 3.b shows an EBIC (Electron Beam Induced Current) picture of the same cell. It is clear that grain boundaries represent the major defects in the material that cause carrier recombination. Figure 4.a shows an EBIC picture- and figure 4.b shows a line scan picture along line AA of a polycrystalline silicon solar cellthat was heat - treated at 850 °C, while figure 4.c is for a line scan BB of the same cell. The deleterious effects of the grain boundaries on the response are clear, causing a sharp reduction in the EBIC signal, whereas twin planes don't show that serious effect. Figure 5 shows an EBIC picture and a line scan picture along line AA of a solar cell that was annealed at 950 °C. Figure 6 shows the effect of annealing on the diade quality factor, n., of the solar cells. In figure 7, the effect of annealing on the minority carrier diffusion<br>length in the base layer of the cell, determined by the surface photovoltage (SPV) method [16] is shown. Figure 8 shows the effect of the annealing condition on the open -  $\epsilon$ circuit voltage and the short - circuit current density of the cells. From this figure it is apparent that annealing at 850 °C is preferable to that at 950 C. Figure 9 shows the effect of annealing on the spectral response of the polycrystalline cells.

VI. Conclusions :

In this work we investigate the effect of heat treatment on polycrystalline silicon thin film solar cells made on acid - treatment upgraded metallurgical grade silicon on graphite by chemical vapor deposition. Heat treatment is<br>found to improve the performance of the solar cells. Further research is required to determine the mechanism of  $\blacksquare$ this improvement and the optimum annealing conditions.

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standard silicon solar cell.

cell under AMl conditions.

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a) EBIC picture





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a) Effect of annealing on  $V_{oc}$  and  $I_{sc}$ 



Figure 8. The effect of annealing on the open-circuit voltage and the short-circuit current density of polycrystalline silicon solar cells.



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